



# Satellite Observations of Tropospheric Ammonia, Methanol, and Formic Acid

Mark W. Shephard, Karen E. Cady-Pereira, Vivienne H. Payne  
AER, Inc.

Ming Luo, Curtis P. Rinsland, Annmarie Eldering, Reinhard Beer, Greg Osterman, Kevin Bowman, Brendan Fisher  
Jet Propulsion Laboratory, California Institute of Technology

Daven Henze  
U. of Colorado

Robert W. Pinder, John Walker  
US EPA

Dylan Millet  
U. of Minnesota



## 1. Motivation

### Ammonia (NH<sub>3</sub>)

Ammonia is an integral part of the nitrogen cycle  
• Nitrogen in ammonia is deposited to the Earth's surface and leads to:  
– Nutrient imbalances  
– Changes in ecosystem composition  
– Algal blooms and hypoxia (oxygen depletion)

### Secondary Inorganic Aerosol Formation

• Ammonia (gas) reacts with sulfate and nitric acid  
– Produces ammonium sulfate and ammonium nitrate (aerosol-phase)  
– Air Quality  
• Responsible for 10-20% of fine particulate matter (PM<sub>2.5</sub>)  
– Exceeds recommended threshold (WHO, 2003)  
• Exposure to aerosol concentration is associated with health issues  
– Cardiovascular disease, inhibited lung development, premature death  
– Climate  
• Direct radiative forcing of ammonium aerosols

### Methanol (CH<sub>3</sub>OH)

Methanol is an important part of the organic carbon cycle  
– The most abundant organic gas in the atmosphere after methane  
– Global emissions are thought to be equal or greater to those of all anthropogenic VOCs combined

### Methanol has important chemical effects in the atmosphere

– It is a major source of formaldehyde in the continental boundary layer  
– It is an important source of CO to the global atmosphere  
– It plays a role in the tropical ozone and HOx budgets (e.g. NOx + VOCs → Ozone)

### Formic Acid (HCOOH)

Formic acid is the most abundant organic acid in the atmosphere, along with acetic acid

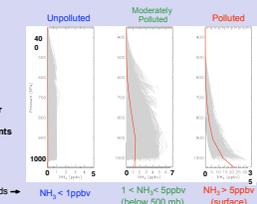
• Formic acid is an important source of free acidity in precipitation "acid rain", especially in remote areas  
• Influences pH-dependent chemical reactions in clouds  
• An important sink for OH(aq) in clouds

### Why Measure from Space?

• Lack of observations to help with large uncertainties in modeled emissions  
• *In situ* (mostly surface) measurements are sparse  
• Uncertainty in the seasonal and spatial variability  
– For example:  
– CMAQ (regional): NH<sub>3</sub> peak emissions during fertilization application in spring (April)  
– GEOS-Chem (global): NH<sub>3</sub> peak emissions with high temperatures in summer (July)

## 3. TES Sensitivity Studies : NH<sub>3</sub> Example

### GEOS-Chem NH<sub>3</sub> Profiles



• Obtained from GEOS-Chem monthly mean output for 2005

• Binned by lower tropospheric ammonia amounts

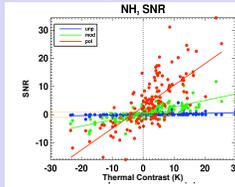
Binning Thresholds →

NH<sub>3</sub> < 1ppbv, 1 < NH<sub>3</sub> < 5ppbv (below 500 mb), NH<sub>3</sub> > 5ppbv (surface)

• Retrieval input profiles for initial guess and a priori  
• Polluted, Moderate, and Unpolluted profiles

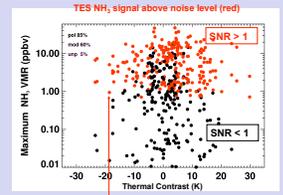
### Sensitivity Test : Strength of NH<sub>3</sub> Signal

• 540 TES L2 profiles + varying amounts of ammonia > FM > simulated spectra



• SNR : (BT<sub>background</sub> - BT<sub>atm</sub>) / NEdT  
• Thermal contrast increases sensitivity  
• Polluted and moderate profiles typically have SNR > 1.  
• The linear fits are used to determine input retrieval profile.

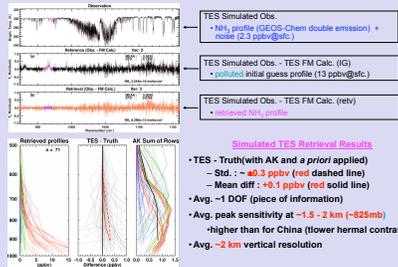
### Sensitivity Test: Level of Detectability



• Detectability level ~1ppbv  
• Conservative level as the:  
– Maximum concentration usually near surface  
– TES most sensitive between 900 and 800 mb

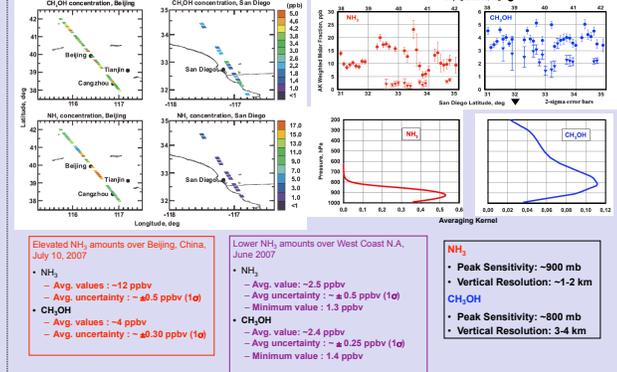
## 4. TES Retrieval Examples

### NH<sub>3</sub> Retrieval Results : TES Simulated Double Emission Profiles



• TES - Truth (with AK and a priori applied)  
– Std. : ~ ±0.3 ppbv (red dashed line)  
– Mean diff. : +0.1 ppbv (red solid line)  
• Avg. ~1 DOF (piece of information)  
• Avg. peak sensitivity at ~1.5 - 2 km (~825mb)  
– higher than for China (lower thermal contrast)  
• Avg. ~2 km vertical resolution

### NH<sub>3</sub> and CH<sub>3</sub>OH TES obs. over Beijing China and West Coast NA

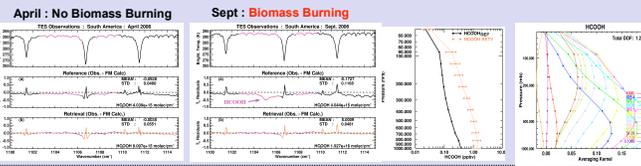


Elevated NH<sub>3</sub> amounts over Beijing, China, July 10, 2007  
• NH<sub>3</sub>  
– Avg. values : ~12 ppbv  
– Avg. uncertainty : ~±0.5 ppbv (1σ)  
• CH<sub>3</sub>OH  
– Avg. values : ~4 ppbv  
– Avg. uncertainty : ~±0.30 ppbv (1σ)

Lower NH<sub>3</sub> amounts over West Coast NA, June 2007  
• NH<sub>3</sub>  
– Avg. values : ~2.5 ppbv  
– Avg. uncertainty : ~±0.5 ppbv (1σ)  
– Minimum value : 1.3 ppbv  
• CH<sub>3</sub>OH  
– Avg. values : ~2.4 ppbv  
– Avg. uncertainty : ~±0.25 ppbv (1σ)  
– Minimum value : 1.4 ppbv

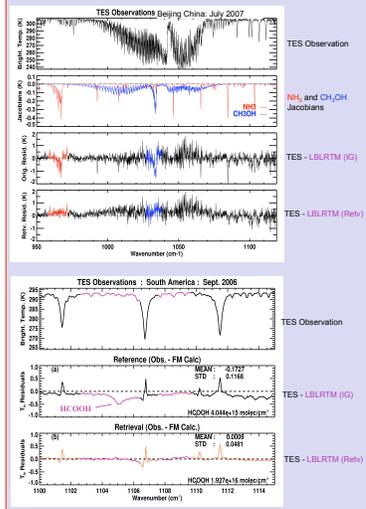
NH<sub>3</sub>  
• Peak Sensitivity : ~900 mb  
• Vertical Resolution : ~1-2 km  
CH<sub>3</sub>OH  
• Peak Sensitivity : ~800 mb  
• Vertical Resolution : 3-4 km

### TES HCOOH Retrieval Example over South America



Elevated TES HCOOH observations associated with Biomass Burning  
• ~1.2 DOFs  
• Peak sensitivity : ~3 - 5 km (~700 - 500mb)  
• Retrieved Values reaching ~1.5 ppbv

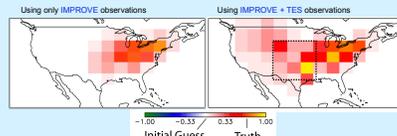
## 2. TES NH<sub>3</sub>, CH<sub>3</sub>OH, HCOOH Spectral Signal



## 5. TES Observations for Emissions: NH<sub>3</sub> Example

### Can TES reduce the large uncertainties in NH<sub>3</sub> emissions?

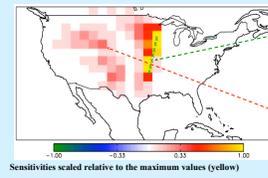
• Inversion modeling to obtain NH<sub>3</sub> model emission scaling factors  
– Input TES NH<sub>3</sub> retrievals from simulated radiance over Midwest.  
– Inversion initial guess emission scaling factors were set to zero  
– Emission scaling factors after 4 iterations of the inversion  
• usually perform more iterations (~20)



• Emission scale factors are moving in the correct direction  
– from initial guess (0.0) towards the double emission truth [ln(2)=0.69]  
• Effects on emissions "upwind" of TES obs.

### How sensitive are TES measurements to changes in NH<sub>3</sub> emissions?

• Show the influence of the NH<sub>3</sub> emissions from any model grid box (from up to a week prior) on the amount of NH<sub>3</sub> that TES observes (marked by X)  
– relative to the influence of the NH<sub>3</sub> directly underneath the TES track



TES Obs are most sensitive to NH<sub>3</sub> emissions directly underneath the track (X)

TES is sensitive to NH<sub>3</sub> emissions away from obs.  
• e.g. TES is ~40% as sensitive to emissions here compared with directly beneath (X)

NH<sub>3</sub> lifetime increased :  
• NH<sub>3</sub> (gas) → NH<sub>3</sub> (aerosol-phase)  
• Bi-directional flux (biosphere and atmosphere)